		STUDY MODULE D	ESCRIPTION FORM			
	f the module/subject struction Chemi	(Code 1010101111010100053			
Field of study			Profile of study (general academic, practical)	Year /Semester		
Sustainable Building Engineering First-cycle			(brak)	1/1		
Elective path/specialty			Subject offered in: Polish	Course (compulsory, elective) obligatory		
Cycle o	f study:		Form of study (full-time,part-time)			
	First-cy	cle studies	full-time			
No. of h	ours			No. of credits		
Lectur	re: 15 Classe	s: - Laboratory: 15	Project/seminars:	. 2		
Status of	of the course in the study	v program (Basic, major, other)	(university-wide, from another fie	ld)		
		(brak)	(brak)			
Educati	on areas and fields of sc	ience and art		ECTS distribution (number and %)		
techr	nical sciences			2 100%		
Resp	onsible for subj	ect / lecturer:	Responsible for subject	/ lecturer:		
dr ir	nż. Agnieszka Ślosarc	zyk	mgr inż. Maria Ratajczak			
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	·	ns of knowledge, skills an	*			
1	Knowledge		atics and chemistry. Knowledge of periodic table and the mpounds (organic and inorganic). Knowledge of basic physical			
	Ability to gain information from the literaure and other sources.					
2	Skills	Ability to write chemical reaction	ns and do the basic stoichiometric calculations.			
3	Social competencies	Awareness of the necessity for o skills.	constant updating and compleme	enting one's knowledge and		
Assu	mptions and ob	jectives of the course:				
	•	e of physicochemical processes oc	curring during production and ap	plication of building materials.		
		omes and reference to the	educational results for a	a field of study		
Knov	vledge:					
		dge of building material choice, de nowledge of building materials res		I and applicable properties, as		
		stands theoretical basics of the ch ction and application - [KSB_W01]	emical and physicochemical pro	cesses occurring in building		
Skills	3:					
	he basis of the gained als - [KSB_U01]	d knowledge the student is able to	characterise the physicochemica	al properties of building		
2. Student is able to adequately choose types of building materials with reference to their practical application - [KSB_U08]						
Socia	al competencies	:				
1. Stuc [KSB_		e the aims during realisation of task	in group, taking into account e.	g. in the social interest -		
2. Stuc	dent has the ability to	plan team work, to divide tasks am orative conclusions (conclusions b				

Assessment methods of study outcomes

Lectures

The exam in the last lecture - the date given at the beginning of the semester. Exam meant to check the ability to write chemical equations, doing basic chemical calculations, and checks the knowledge of basic physicochemical properties of building materials.

Laboratory classes

A short verbal test at the beginning of the class. A colloquium in the form of open and closed questionsat the end of the semester covering the material of the laboratory classes.

Grade scale:

100-90% of maximum possible points - 5.0

90-80% of maximum possible points - 4.5

80-70% of maximum possible points - 4.0

70-60% of maximum possible points - 3.5

60-50% of maximum possible points - 3.0

Course description

Lecture

Chemistry versus construction. Chemical reactions in aqueous environment.

Complex systems occurring in construction; colloidal systems.

Types of chemical compounds and chemical processes occurring during production, application and exploitation of building materials.

Chemical composition and structure of building materials as determinants of their physicomechanical and applicable properties. Thermodynamic conditions of durability of building materials. Phase transitions.

Basics of crystal chemistry of building materials. The structure of silicates and aluminosilicate minerals.

Kinetics of chemical reactions occurring in construction. Catalysis.

Chemistry of mineral binders. Hydraulic and air binders. Processes occurring during obtaining, bonding and hardening of cement, lime, gypsum, silicate and magnesium binders.

Structure and properties of metals applied in construction.

Polymers as components of plastics used in construction, their properties and obtaining.

Processes occurring during degradation of building materials. Corrosion of concrete. Corrosion of reinforcing bars in reinforced concrete. Corrosion of polymers and glass.

Topics of the laboratory classes:

1. Basics of the chemical quality analysis. Identification analysis of chosen cations.

2. Basics of the chemical quantity analysis. Defining the sodium hydroxide concentration with the use of the acid-base titration.

3. Hydrolysis of the salts and defining the pH of the aqueous solutions.

4. Kinetics of chemical reactions.

5. Corrosion of building materials. Estimation of the corrosion stage of cement stone and definition of the stage of concrete carbonisation.

6. Chemical corrosion of steel. Estimation of the corrosion strength of ordinary steel and that with anti-corrosive coating.

Basic bibliography:

1. R.M.E. Diamant, Chemistry of building materials.

Additional bibliography:

Result of average student's workload

Activity	Time (working hours)
1. Participation in the lectures	15
2. Participation in the laboratory classes	15
3. Preparation to the laboratory classes	5
4. Preparation to the colloquium at the laboratory classes	5
5. Preparation to the examin	10

Student's workload				
Source of workload	hours	ECTS		
Total workload	50	2		
Contact hours	30	1		
Practical activities	15	1		